6376 measured reflections

 $R_{\rm int} = 0.035$ 

2335 independent reflections

1951 reflections with  $I > 2\sigma(I)$ 

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## 3,5-Dimethyl-1-phenyl-1*H*-pyrazole-4carbaldehyde

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Key indicators: single-crystal X-ray study; T = 100 K; mean  $\sigma$ (C–C) = 0.005 Å; R factor = 0.080; wR factor = 0.194; data-to-parameter ratio = 16.9.

In the title molecule,  $C_{12}H_{12}N_2O$ , the five- and six-membered rings form a dihedral angle of 68.41 (16)°. The aldehyde group is nearly coplanar with the pyrazole ring [C-C-C-O torsion angle = -0.4 (5)°]. The three-dimensional architecture is sustained by weak  $C-H\cdots O$  and  $C-H\cdots \pi$  interactions.

#### **Related literature**

For the anti-bacterial properties of pyrazole derivatives, see: Kane *et al.* (2003). For related structures, see: Asiri *et al.* (2012a,b).



#### Experimental

 $\begin{array}{l} Crystal \ data \\ C_{12}H_{12}N_2O \\ M_r = 200.24 \\ Monoclinic, \ P2_1/c \\ a = 6.6264 \ (4) \ \AA \\ b = 6.7497 \ (4) \ \AA \\ c = 22.6203 \ (12) \ \AA \\ \beta = 94.785 \ (5)^\circ \end{array}$ 

 $V = 1008.19 (10) Å^{3}$ Z = 4 Mo K\alpha radiation  $\mu = 0.09 \text{ mm}^{-1}$ T = 100 K 0.25 \times 0.15 \times 0.05 mm

#### Data collection

```
Agilent SuperNova Dual
diffractometer with an Atlas
detector
Absorption correction: multi-scan
(CrysAlis PRO; Agilent, 2011)
T_{\rm min} = 0.979, T_{\rm max} = 0.996
```

#### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.080$ 138 parameters $wR(F^2) = 0.194$ H-atom parameters constrainedS = 1.23 $\Delta \rho_{max} = 0.43 \text{ e } \text{\AA}^{-3}$ 2335 reflections $\Delta \rho_{min} = -0.31 \text{ e } \text{\AA}^{-3}$ 

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C7-C12 ring.

$D - H \cdot \cdot \cdot A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} \hline C8-H8\cdots O1^{i} \\ C11-H11\cdots Cg1^{ii} \end{array}$	0.95 0.95	2.43 2.71	3.315 (4) 3.509 (4)	155 142
Symmetry codes: (i) -	x + 1, -y + 1, -	-z + 1; (ii) $-x$ ,	$y - \frac{1}{2}, -z + \frac{1}{2}.$	

 $\frac{1}{2} = \frac{1}{2} = \frac{1}$ 

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5480).

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# supplementary materials

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# 3,5-Dimethyl-1-phenyl-1H-pyrazole-4-carbaldehyde

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#### Comment

In continuation of structural studies of pyrazole derivatives (Asiri *et al.*, 2012*a*; Asiri *et al.*, 2012*b*), motivated by their putative biological activity (Kane *et al.*, 2003), the title compound, 3,5-dimethyl-1-phenyl-1*H*-4-pyrazole-3-carboxaldehyde (I), was investigated crystallographically.

In (I), Fig. 1, there is a twist about the single bond linking the five- and six-membered rings with the N2—N1—C7—C8 torsion angle being -112.1 (3)  $^{\circ}$ ; the dihedral angle between the rings is 68.41 (16)  $^{\circ}$ . The aldehyde group is co-planar with the pyrazole ring to which it is connected as seen in the value of the C2—C3—C6—O1 torsion angle of -0.4 (5) $^{\circ}$ .

Molecules are connected into the three-dimensional architecture by C—H···O and C—H··· $\pi$  interactions, Fig. 2 and Table 1.

#### Experimental

To a cold solution of N,N-dimethylformamide (1.46 g, 20 mmol), freshly distilled phosphorous oxychloride (1.54 g, 10 mmol) was added with stirring over a period of 30 min. A solution of 3,5-dimethyl-1-phenyl-1*H*-4-pyrazole-3- carboxaldehyde (1.72 g, 10 mmol) in N,N-dimethylformamide (15 ml) was added drop-wise while maintaining the temperature between 273–278 K. The resulting mixture was heated under reflux for 1 h, cooled and poured with continuous stirring into crushed ice. After 15 min, the precipitate was filtered and crystallized from aqueous ethanol to give needles. Yield: 69%. *M*.pt: 397–399 K.

#### Refinement

Carbon-bound H-atoms were placed in calculated positions [C—H = 0.95 to 0.98 Å,  $U_{iso}$ (H) = 1.2 to 1.5 $U_{eq}$ (C)] and were included in the refinement in the riding model approximation. Owing to poor agreement, the (2 1 0) reflection was omitted from the final cycles of refinement.

#### **Computing details**

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO* (Agilent, 2011); data reduction: *CrysAlis PRO* (Agilent, 2011); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).



#### Figure 1

The molecular structure of (I) showing the atom-labelling scheme and displacement ellipsoids at the 50% probability level.



#### Figure 2

A view in projection down the *a* axis of the unit-cell contents of (I). The C—H···O and C—H··· $\pi$  interactions are shown as orange and purple dashed lines, respectively.

#### 3,5-Dimethyl-1-phenyl-1*H*-pyrazole-4-carbaldehyde

Crystal data	
$C_{12}H_{12}N_2O$	F(000) = 424
$M_r = 200.24$	$D_{\rm x} = 1.319 {\rm Mg m^{-3}}$
Monoclinic, $P2_1/c$	Mo <i>K</i> $\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: -P 2ybc	Cell parameters from 2660 reflections
a = 6.6264 (4)  Å	$\theta = 2.7 - 27.5^{\circ}$
b = 6.7497 (4) Å	$\mu = 0.09 \text{ mm}^{-1}$
c = 22.6203 (12) Å	T = 100  K
$\beta = 94.785(5)^{\circ}$	Prism, colourless
$V = 1008.19 (10) Å^3$	$0.25 \times 0.15 \times 0.05 \text{ mm}$
Z = 4	

Data collection

Agilent SuperNova Dual	$T_{\min} = 0.979, T_{\max} = 0.996$
diffractometer with an Atlas detector	6376 measured reflections
Radiation source: SuperNova (Mo) X-ray	2335 independent reflections
Source	1951 reflections with $I > 2\sigma(I)$
Mirror monochromator	$R_{\text{int}} = 0.035$
Detector resolution: 10.4041 pixels mm <sup>-1</sup>	$\theta_{\text{max}} = 27.6^{\circ}, \theta_{\text{min}} = 3.1^{\circ}$
$\omega$ scan	$h = -8 \rightarrow 8$
Absorption correction: multi-scan	$k = -8 \rightarrow 6$
( <i>CrysAlis PRO</i> ; Agilent, 2011)	$l = -29 \rightarrow 29$
Refinement	
Refinement on $F^2$	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.080$	Hydrogen site location: inferred from
$wR(F^2) = 0.194$	neighbouring sites
S = 1.23	H-atom parameters constrained
2335 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0317P)^2 + 2.9806P]$
138 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{max} = 0.001$
Primary atom site location: structure-invariant	$\Delta\rho_{max} = 0.43$ e Å <sup>-3</sup>
direct methods	$\Delta\rho_{min} = -0.31$ e Å <sup>-3</sup>

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(Å^2)$ 

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
01	0.2480 (3)	0.6126 (4)	0.58330 (9)	0.0215 (5)
N1	0.2395 (4)	0.2683 (4)	0.40971 (10)	0.0148 (5)
N2	0.2307 (4)	0.1335 (4)	0.45524 (11)	0.0195 (6)
C1	0.2332 (6)	0.1451 (5)	0.56361 (14)	0.0276 (8)
H1A	0.1190	0.0526	0.5626	0.041*
H1B	0.2177	0.2463	0.5940	0.041*
H1C	0.3600	0.0727	0.5730	0.041*
C2	0.2378 (5)	0.2431 (5)	0.50395 (13)	0.0172 (6)
C3	0.2500 (4)	0.4477 (5)	0.49014 (13)	0.0148 (6)
C4	0.2499 (4)	0.4563 (5)	0.42869 (13)	0.0153 (6)
C5	0.2611 (6)	0.6283 (5)	0.38750 (14)	0.0236 (7)
H5A	0.1524	0.6179	0.3555	0.035*
H5B	0.3925	0.6280	0.3706	0.035*
H5C	0.2456	0.7519	0.4094	0.035*
C6	0.2550 (4)	0.6179 (5)	0.52932 (13)	0.0167 (6)
H6	0.2644	0.7448	0.5116	0.020*
C7	0.2304 (4)	0.1953 (4)	0.34975 (12)	0.0143 (6)
C8	0.4000 (5)	0.2103 (5)	0.31798 (14)	0.0208 (7)
H8	0.5214	0.2683	0.3354	0.025*
C9	0.3893 (5)	0.1389 (5)	0.25999 (14)	0.0235 (7)
H9	0.5033	0.1502	0.2374	0.028*
C10	0.2120 (5)	0.0509 (5)	0.23500 (13)	0.0203 (7)
H10	0.2060	0.0009	0.1956	0.024*
C11	0.0447 (5)	0.0363 (5)	0.26753 (13)	0.0182 (6)
H11	-0.0756	-0.0246	0.2505	0.022*
C12	0.0520 (5)	0.1106 (4)	0.32508 (13)	0.0165 (6)

# supplementary materials

H12	-0.0636	0.1	034	0.3472	0.020*	
Atomic	displacement para	ameters ( $Å^2$ )				
	$U^{11}$	U <sup>22</sup>	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
01	0.0239 (11)	0.0255 (12)	0.0150 (10)	-0.0002 (10)	0.0007 (8)	-0.0037 (9)
N1	0.0210 (12)	0.0131 (12)	0.0105 (11)	-0.0012 (10)	0.0024 (9)	0.0006 (9)
N2	0.0305 (14)	0.0157 (13)	0.0126 (12)	-0.0021 (11)	0.0031 (10)	0.0009 (10)
C1	0.048 (2)	0.0210 (17)	0.0140 (15)	-0.0048 (16)	0.0052 (14)	0.0023 (13)
C2	0.0219 (15)	0.0166 (14)	0.0134 (14)	-0.0009 (12)	0.0020 (11)	0.0008 (11)
C3	0.0144 (13)	0.0171 (14)	0.0133 (13)	-0.0012 (12)	0.0031 (10)	-0.0005 (11)
C4	0.0157 (13)	0.0157 (15)	0.0140 (14)	-0.0013 (12)	-0.0020 (11)	0.0000 (11)
C5	0.0401 (19)	0.0145 (15)	0.0160 (15)	-0.0003 (15)	0.0016 (13)	-0.0004 (12)
C6	0.0170 (14)	0.0173 (15)	0.0157 (14)	-0.0012 (12)	0.0012 (11)	-0.0024 (12)
C7	0.0218 (14)	0.0092 (13)	0.0118 (13)	0.0006 (11)	0.0004 (11)	0.0001 (10)
C8	0.0211 (15)	0.0238 (16)	0.0175 (15)	-0.0033 (13)	0.0017 (12)	-0.0043 (13)
C9	0.0230 (15)	0.0280 (18)	0.0205 (15)	-0.0003 (14)	0.0075 (12)	-0.0071 (14)
C10	0.0298 (17)	0.0183 (15)	0.0129 (14)	0.0024 (13)	0.0012 (12)	-0.0031 (12)
C11	0.0242 (15)	0.0139 (14)	0.0155 (14)	-0.0013 (12)	-0.0038 (11)	-0.0001 (12)
C12	0.0206 (14)	0.0140 (14)	0.0149 (13)	-0.0016 (12)	0.0024 (11)	0.0008 (11)

## Geometric parameters (Å, °)

01—C6	1.226 (4)	С5—Н5В	0.9800
N1—C4	1.339 (4)	С5—Н5С	0.9800
N1—N2	1.379 (3)	С6—Н6	0.9500
N1—C7	1.439 (4)	C7—C12	1.387 (4)
N2—C2	1.325 (4)	C7—C8	1.388 (4)
C1—C2	1.506 (4)	C8—C9	1.394 (4)
C1—H1A	0.9800	C8—H8	0.9500
C1—H1B	0.9800	C9—C10	1.393 (5)
C1—H1C	0.9800	С9—Н9	0.9500
C2—C3	1.419 (4)	C10—C11	1.385 (4)
C3—C4	1.391 (4)	C10—H10	0.9500
C3—C6	1.450 (4)	C11—C12	1.392 (4)
C4—C5	1.494 (4)	C11—H11	0.9500
С5—Н5А	0.9800	C12—H12	0.9500
C4—N1—N2	112.9 (2)	H5A—C5—H5C	109.5
C4—N1—C7	128.6 (2)	H5B—C5—H5C	109.5
N2—N1—C7	118.5 (2)	O1—C6—C3	125.8 (3)
C2—N2—N1	104.6 (2)	O1—C6—H6	117.1
C2—C1—H1A	109.5	С3—С6—Н6	117.1
C2—C1—H1B	109.5	C12—C7—C8	121.4 (3)
H1A—C1—H1B	109.5	C12—C7—N1	119.2 (3)
C2—C1—H1C	109.5	C8—C7—N1	119.4 (3)
H1A—C1—H1C	109.5	C7—C8—C9	118.8 (3)
H1B—C1—H1C	109.5	С7—С8—Н8	120.6
N2—C2—C3	111.0 (3)	С9—С8—Н8	120.6
N2—C2—C1	119.9 (3)	C10—C9—C8	120.3 (3)

C3—C2—C1	129.1 (3)	С10—С9—Н9	119.9
C4—C3—C2	105.4 (3)	С8—С9—Н9	119.9
C4—C3—C6	125.2 (3)	C11—C10—C9	120.0 (3)
C2—C3—C6	129.4 (3)	C11—C10—H10	120.0
N1—C4—C3	106.1 (3)	С9—С10—Н10	120.0
N1—C4—C5	122.7 (3)	C10-C11-C12	120.3 (3)
C3—C4—C5	131.3 (3)	C10-C11-H11	119.8
С4—С5—Н5А	109.5	C12—C11—H11	119.8
C4—C5—H5B	109.5	C7—C12—C11	119.1 (3)
H5A—C5—H5B	109.5	С7—С12—Н12	120.4
C4—C5—H5C	109.5	C11—C12—H12	120.4
C4—N1—N2—C2	-0.6 (3)	C6—C3—C4—C5	2.2 (5)
C7—N1—N2—C2	-178.7 (3)	C4—C3—C6—O1	177.3 (3)
N1—N2—C2—C3	0.3 (3)	C2-C3-C6-O1	-0.4 (5)
N1—N2—C2—C1	-179.5 (3)	C4—N1—C7—C12	-110.1 (4)
N2-C2-C3-C4	0.0 (4)	N2—N1—C7—C12	67.7 (4)
C1—C2—C3—C4	179.8 (3)	C4—N1—C7—C8	70.2 (4)
N2-C2-C3-C6	178.1 (3)	N2—N1—C7—C8	-112.1 (3)
C1—C2—C3—C6	-2.2 (6)	C12—C7—C8—C9	0.2 (5)
N2—N1—C4—C3	0.6 (3)	N1—C7—C8—C9	179.9 (3)
C7—N1—C4—C3	178.5 (3)	C7—C8—C9—C10	-1.1 (5)
N2—N1—C4—C5	179.9 (3)	C8—C9—C10—C11	0.8 (5)
C7—N1—C4—C5	-2.2 (5)	C9—C10—C11—C12	0.5 (5)
C2-C3-C4-N1	-0.3 (3)	C8—C7—C12—C11	1.1 (5)
C6—C3—C4—N1	-178.5 (3)	N1-C7-C12-C11	-178.7 (3)
C2—C3—C4—C5	-179.6 (3)	C10-C11-C12-C7	-1.4 (5)

### Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C7–C12 ring.

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H··· $A$
C8—H8····O1 <sup>i</sup>	0.95	2.43	3.315 (4)	155
C11—H11…Cg1 <sup>ii</sup>	0.95	2.71	3.509 (4)	142

Symmetry codes: (i) -*x*+1, -*y*+1, -*z*+1; (ii) -*x*, *y*-1/2, -*z*+1/2.